

## Modelisation of the contribution of the Na/Ca exchanger to cell membrane potential and intracellular ion concentrations

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**Abstract.** Modelisation plays a significant role in the study of ion transfer through the cell membrane and in the comprehension of cellular excitability. We were interested in the selective ion transfers through the  $K_{Ca}$ ,  $Na_v$ ,  $Ca_v$  channels and the Na/Ca exchanger (NCX). The membrane behaves like an electric circuit because of the existence of ion gradients maintained by the cell. The non-linearity of this circuit gives rise to complex oscillations of the membrane potential. By application of the finite difference method (FDM) and the concept of percolation we studied the role of the NCX in the regulation of the intracellular  $Ca^{2+}$  concentration and the oscillations of the membrane potential. The fractal representation of the distribution of active channels allows us to follow the diffusion of intracellular  $Ca^{2+}$  ions. These calculations show that the hyperpolarization and the change in the burst duration of the membrane potential are primarily due to the NCX.

**Key words:** Percolation — Finite difference method — Ion channels and concentrations — Current clamp — Voltage clamp

### Introduction

The plasma membrane constitutes a selective barrier between the inside and the outside of a cell. It controls the entry and the exit of various molecules and ions between the two compartments. All cells develop a potential difference (PD) between the two faces of the membrane. This PD is responsible for the electric activity of excitable cells, which appears in the form of oscillations of the membrane potential. These oscillations take place in many physiological processes, in particular in neuronal, cardiac and pancreatic cells (Wakimoto et al. 2000; Barg et al. 2001; Munekazu and Iwamoto 2001; Bano et al. 2005).

The aim of this work was to evaluate the role of a membrane protein, the Na/Ca exchanger (NCX) in the regulation of the oscillations of the membrane potential. NCX allows  $Ca^{2+}$  extrusion from the cell and entry of  $Na^+$  along its electrochemical gradient without energy consumption. In addition, because NCX is electrogenic and voltage dependent, it can reverse dur-

ing cellular activation and contributes to  $Ca^{2+}$  entry into the cell (Blaustein and Lederer 1999). In  $\beta$ -cells and the heart, NCX seems to be the predominant mechanism for  $Ca^{2+}$  extrusion, accounting for approximately 70 and 90% of  $Ca^{2+}$  extrusion, respectively (Bers et al. 1996; Van Eyllen et al. 1998).

We used the percolation concept introduced in 1953 by Hammersley in order to describe statistical systems made up of a great number of objects which can be connected between them. According to the number of objects put in contact, the long-range communication is either possible or non-existent. Between the two modes of communication, there is a threshold of precise transition, called percolation threshold (Stauffer and Aharony 1991). The critical behaviour of the system in the vicinity of this threshold is characteristic of a phase transition.

In the case of biological systems, the research of the transport properties is a complex problem due to its mathematical aspect. It is more judicious to study the biological systems by modelling them on simple networks. By using the two-dimensional networks, we can simulate many systems and have very approximate values of the parameters, which characterise them.

The first computational model of the action potential (AP) was formulated by Hodgkin and Huxley for the

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axon (Hodgkin 1948; Hodgkin and Huxley 1952). Their circuit model of the cell membrane remains the basis for modern AP models. In this case, we modelled the transfer of ions through channels of a cellular membrane by using a square electrical network of conductances randomly distributed, which represent three different channels: the potassium channels ( $K_{Ca}$ ), activated by intracellular  $Ca^{2+}$ , the voltage-gated sodium channels ( $Na_v$ ) and the voltage gated calcium channels ( $Ca_v$ ). The fuses are connected in series with the conductances to highlight the activation and the inactivation of channels. The dielectric property of the phospholipids is represented by a capacity in parallel with the network. The phase transition of the system is described by the change in the vicinity of the percolation threshold from the non-permeability to the permeability of the membrane.

Our goal was to study the diffusion of ions through the membrane. For that, we have to use the Fick's laws, which treat partial derivative equations. Analytically, the solution of these equations is very difficult and sometimes impossible. We used a very simple numerical method – the finite difference method (FDM) (Garrido et al. 1985; Morton and Mayers 1995; Miloshevsky et al. 2006) in order to find the concentrations at each node of the network. FDM proceeds by replacing the derivatives of the differential equations by finite difference approximations. This gives a large algebraic system of equations, which can be solved in place of the differential equations.

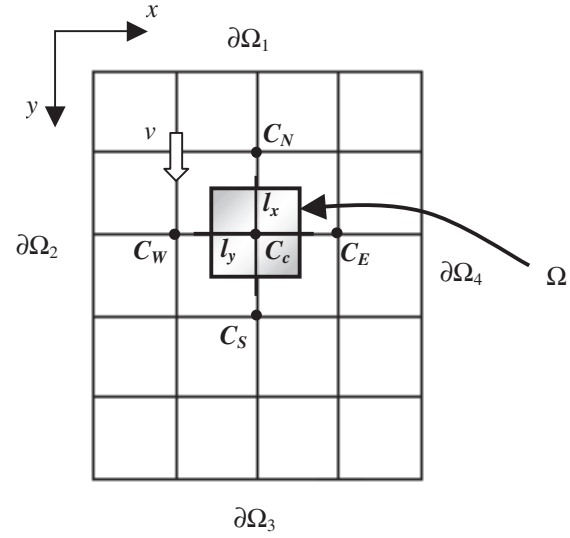
We used to simulate two measurement techniques: the current-clamp (Hilgemann 1988; Zhan et al. 1999; Dallas et al. 2008) and the voltage-clamp (González-Caballero et al. 1988; Leois and Rae 1998) for better including the ion mechanisms. This model enabled us to study the effect of NCX on the membrane potential and the change in intracellular  $Ca^{2+}$  concentration.

#### *Ion transfer and FDM*

In our simulation, we have to solve partial derivative equations. For that, it is necessary to use a numerical method of resolution, we chose FDM for its mathematical and data-processing simplicity.

The transfer of the ions through the channels is a nonstationary electrochemical diffusion (Bergamini et al. 1998). We consider the regular mesh of field  $\Omega$  represented in Fig. 1. We study a problem of advection diffusion (Hundsdoerfer and Verwer 2003; El Makrini et al. 2007) characterised by a uniform transport velocity  $v$  according to the  $x$  direction and the diffusion coefficients  $D_\alpha$  ( $\alpha = K^+, Na^+, Ca^{2+}$ , constant for each ion).

The equation controlling the ion transport is given by the Fick's 2<sup>nd</sup> law (Flynn 1972; Flynn et al. 1974; Rappaz et al. 1998):



**Figure 1.** Representation of the regular mesh of field  $\Omega$  of dimension  $l_x = l_y = l$ . E (East), W (West), N (North), S (South) are the neighbours of the central node  $C$ .

$$\frac{\partial C_\alpha}{\partial t} - D_\alpha \operatorname{div}(\operatorname{grad} C_\alpha) + v \operatorname{grad} C_\alpha = \frac{\partial C_\alpha}{\partial t} - D_\alpha \left[ \frac{\partial^2 C_\alpha}{\partial x^2} + \frac{\partial^2 C_\alpha}{\partial y^2} \right] + v \frac{\partial C_\alpha}{\partial x} = 0 \quad (1)$$

With  $C_\alpha(x, y, t)$  are the concentrations of  $K^+$ ,  $Na^+$ ,  $Ca^{2+}$ , respectively.

To solve this equation, we take account of the boundary conditions on the four borders (Rappaz et al. 1998):

$$\partial\Omega_1 : C_\alpha(x, t) = \tilde{C}_\alpha \quad (\text{Dirichlet condition}) \quad (2)$$

$$\partial\Omega_2 : D \frac{\partial C_\alpha}{\partial n} = -D \frac{\partial C_\alpha}{\partial y} = q \quad (\text{Neumann condition, imposed flow}) \quad (3)$$

$$\partial\Omega_3 : D \frac{\partial C_\alpha}{\partial n} = -D \frac{\partial C_\alpha}{\partial x} = 0 \quad (\text{homogeneous Neumann condition}) \quad (4)$$

$$\partial\Omega_4 : D \frac{\partial C_\alpha}{\partial n} = -D \frac{\partial C_\alpha}{\partial y} = -\beta(C_\alpha - \tilde{C}_\alpha) \quad (\text{Cauchy condition}) \quad (5)$$

To apply the FDM, the rectangular field of dimension  $l_x = l_y = l$  is squared according to the axes directions as illustrated in Fig. 1. The neighbours of a central node of index  $C$  are located by the letters E (east), W (west), N (north), S (south). If the neighbour exists, we call  $h$  the distance separating it from the central node.

By using the Taylor's series expansions and the FDM at the node  $C$ , the Eq. (1) becomes:

$$\frac{1}{F_{Oh}}(C_{ac}^{j+1} - C_{ac}^j) - (C_{ae} + C_{aw} + C_{av} + C_{as} - 4C_{ac})^\theta + \frac{P_{eh}}{2}(C_{ae} - C_{aw})^\theta = 0 \quad (6)$$

The linear combinations of the Eq. (6) are indexed with  $\theta$  to separate the explicit and implicit finite difference forms.  $P_{eh}$  is a Péclet number:  $P_{eh} = (v \cdot h)/D$ ,  $F_{Oh}$  is a Fourier number:  $F_{Oh} = (D \cdot \Delta t)/h^2$ .

By using the Courant, Friedrichs and Levy condition (CFL) which is  $D \cdot \Delta t/h^2 \leq 1/4$  at  $v \cdot h/D \leq 1/2$ , we impose the stability criteria  $F_{Oh} \leq 1/4$  at  $P_{eh} \leq 2$  (Strikwerda 1989; Rappaz et al. 1998).

### The model

We keep the same structure of grid like the FDM and we use a square network of size  $L$ . The conductive ion channels are characterised by their elementary conductance:  $\gamma_{K-Ca}$ ,  $\gamma_{Na}$ ,  $\gamma_{Ca}$  randomly distributed with open probability:  $P_{K-Ca}$ ,  $P_{Na}$ , and  $P_{Ca}$ , and with intracellular ion concentrations  $C_{K-Ca}$ ,  $C_{Na}$  and  $C_{Ca}$ , respectively. The probability of opening and closing of ion channels is represented by a fuse connected in series with the conductances. The dielectric character of the double-layer of lipid is represented by a capacity  $C_m$  (Fig. 2).

We chose three types of ion channels:

- The potassium channels ( $K_{Ca}$ ), activated by intracellular  $Ca^{2+}$  (Romero et al. 1998; Siwy et al. 2001; Ledoux et al. 2006; Zhao et al. 2007), named BK or maxi-K slightly sensitive to the variations of the membrane potential whose conductance varies between 180 and 300 pS.
- The voltage-gated sodium channels ( $Na_v$ ), responsible of the ascending phase of the action potential (Stuart and

Hausser 1994; Hollerbach et al. 2000), with a conductance of 10–15 pS.

- The voltage gated calcium channels ( $Ca_v$ ) whose conductance varies between 11 and 20 pS (Nonner and Eisenberg 1998; Boda et al. 2004; Elbasiouny et al. 2005).

The current through a selective channel is given by the Ohm's law and the Nernst equation (Hille 2001). We apply to the circuit the first Kirchhoff law:

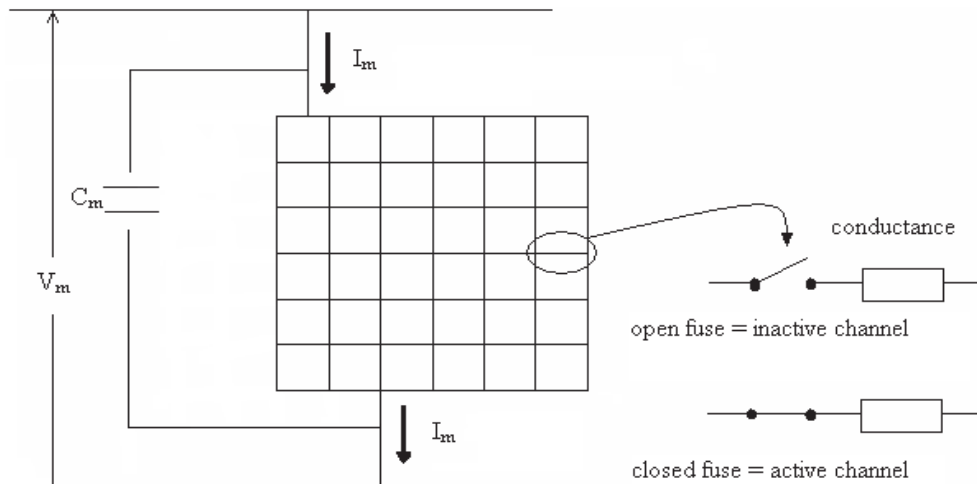
$$C_m \frac{dV}{dt} = -I_{ion}(V) \quad (7)$$

In our model, the total current ( $I_{ion}$ ) is the sum of three currents:  $I_{Na}$  which represents the depolarizing  $Na^+$  current,  $I_{K-Ca}$  that accounts for the hyperpolarizing  $K^+$  current and the slowly activating  $Ca^{2+}$  current  $I_{Ca}$ . We shall assume for simplicity that: i) the relaxation kinetics are first order (Milescu et al. 2005; Rudy and Silva 2006) and described by any time homogeneous Markov process (Goldman 1991; Venkataramanan and Sigworth 2002; Faber et al. 2007), in which the channel jumps from the open state to the closed state; ii) the time constants of the  $Ca^{2+}$  and  $Na^+$  current are small compared to the potential bursting, for that we use the measured values for parameters and theoretical expressions for activation levels given by Rorsman (Rorsman and Trube 1986) and Sherman (Sherman et al. 1988).

These currents are given according to the Hodgkin-Huxley (Hodgkin and Huxley 1952) and Morris-Lecar models (Morris and Lecar 1981; Rinzel and Ermentrout 1999), and have the following expressions:

$$I_{Na} = \gamma_{Na} \cdot m^3 \cdot h \cdot (V_m - E_{Na}) \quad (8)$$

$$I_{K-Ca} = \gamma_{K-Ca} \cdot z \cdot (V_m - E_K) \quad (9)$$



**Figure 2.** Schematization of the square network of size  $6 \times 6$  of conductances and fuses randomly distributed.

$$I_{Ca} = \gamma_{Ca} \cdot m \cdot h \cdot (V_m - E_{Ca}) \quad (10)$$

where:  $E_{Na}$ ,  $E_K$  and  $E_{Ca}$  are the reversal potentials (in mV) and  $m$  is the open probability of the activation gate, described by

$$m = \frac{1}{1 + \exp[(4 - V)/14]} \quad (11)$$

$h$  is the open probability of a single first-order inactivation gate, given by the sigmoidal function:

$$h = \frac{1}{1 + \exp[(V + 10)/10]} \quad (12)$$

$z$  is the gating variable with a Hill-like dependence on  $C_{Ca}$

$$z = \frac{C_{Ca}}{C_{Ca} + 1} \quad (13)$$

and the balance equation for  $C_{Ca}$  is:

$$\frac{dC_{Ca}}{dt} = 0.005(-0.2 \cdot I_{Ca} - C_{Ca}) \quad (14)$$

The FDM imposes known concentrations on the four edges of the circuit. We chose the potassium concentrations  $C_{K-Ca}$  because, at rest, the membrane is essentially permeable to potassium.

## Results and Discussion

In our model, the elementary conductances are:  $\gamma_K = 200$  pS,  $\gamma_{Na} = 12$  pS and  $\gamma_{Ca} = 15$  pS. The membrane capacity per unit of area of the biological membranes is:  $C_m = 1$   $\mu$ F/cm<sup>2</sup>.

The intracellular concentrations have values higher or equal to physiological values such as:  $C_{K-Ca} = 100$  mmol/l,  $C_{Na} = 10$  mmol/l and  $C_{Ca} = 1$   $\mu$ mol/l. The extracellular concentrations will not be modified and correspond to the physiological values (Hille 2001).

Since the studied problem is a pure diffusion, the numbers of Fourier and Péclet are  $F_{Oh} = 1/2$  and  $P_{eh} = 1$  (Phannkuch 1963; Muradoglu and Tryggvason 2008).

To solve the differential Eqs. (7) and (14) we used the numerical algorithms group library of Fortran and the D02BBF subroutine (Ermentrout 2003; Metcalf et al. 2004).

The direction of the ion flows, the gradients of concentration and the variations of membrane potential are controlled for any iteration. The results are obtained after 1000 iterations.

### Percolation threshold

In a random system, the transport phenomena are studied in the vicinity of the percolation threshold. We have to find the threshold probability for which the membrane forwards

impermeable phase (non-conducting) to the permeable phase (conducting).

We varied the probabilities  $P_K$ ,  $P_{Na}$  and  $P_{Ca}$  from 0 to 1, for the intracellular concentrations  $C_{K-Ca} = 140$  mmol/l,  $C_{Na} = 20$  mmol/l and  $C_{Ca} = 1$   $\mu$ mol/l. For each probability, we calculated the membrane conductance by the star-triangle transformation, in order to reduce the matrix size from  $[L^2 \times L^2]$  to  $[L \times (L + 1)]$ :

$$v_e^{(\chi)}(i,j) = [h_o(i,j)^{(\chi-1)} \times v_e^{(\chi-1)}(i,j) + v_e^{(\chi-1)}(i+1,j) \times v_e^{(\chi-1)}(i,j) + v_e^{(\chi-1)}(i+1,j) \times h_o^{(\chi-1)}(i,j)] / v_e^{(\chi-1)}(i,j) \quad (9)$$

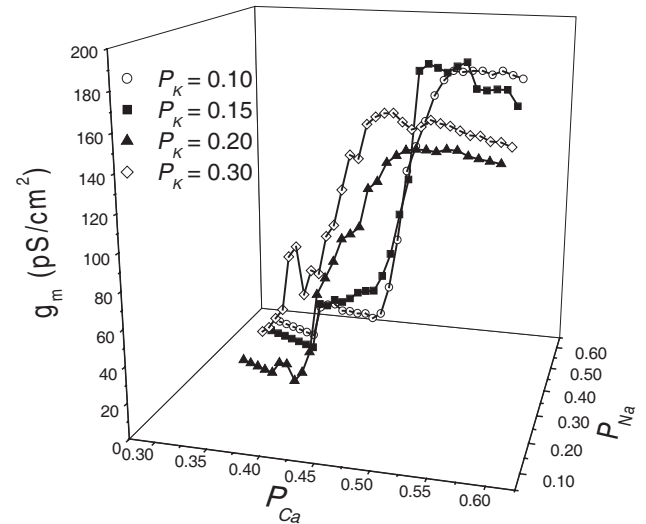
$$v_e^{(\chi)}(i+1,j) = [h_o(i,j)^{(\chi-1)} \times v_e^{(\chi-1)}(i,j) + v_e^{(\chi-1)}(i+1,j) \times v_e^{(\chi-1)}(i,j) + v_e^{(\chi-1)}(i+1,j) \times h_o^{(\chi-1)}(i,j)] / h_o^{(\chi-1)}(i,j) \quad (10)$$

$$h_o^{(\chi)}(i,j) = [h_o(i,j)^{(\chi-1)} \times v_e^{(\chi-1)}(i,j) + v_e^{(\chi-1)}(i+1,j) \times v_e^{(\chi-1)}(i,j) + v_e^{(\chi-1)}(i+1,j) \times h_o^{(\chi-1)}(i,j)] / v_e^{(\chi-1)}(i+1,j) \quad (11)$$

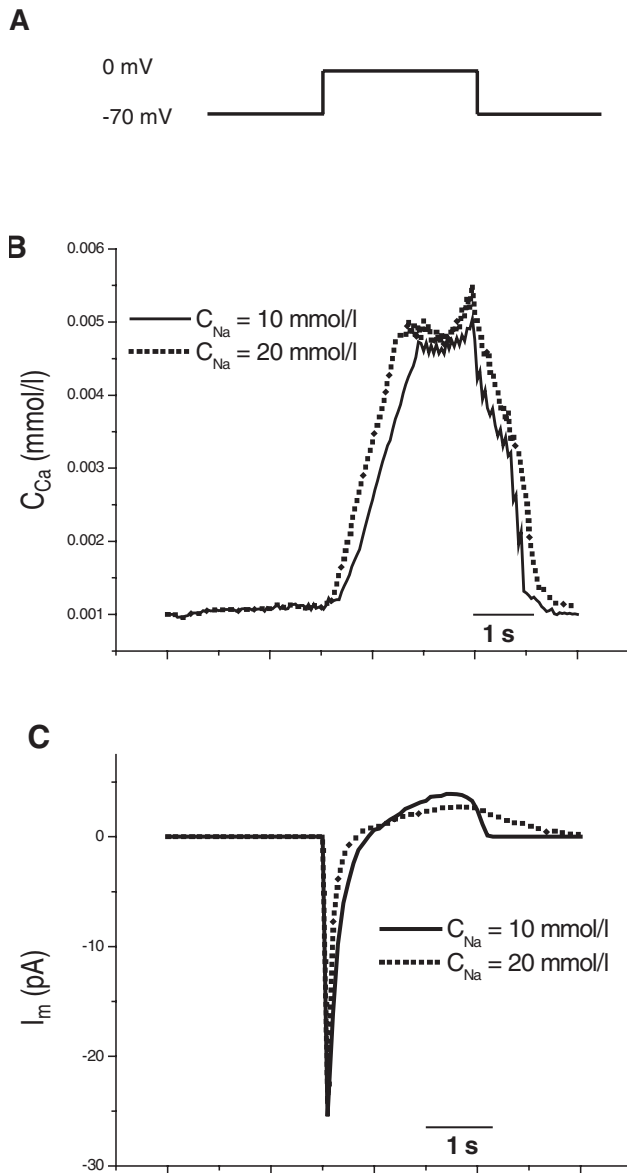
Where  $v_e(i,j)$  and  $h_o(i,j)$  represent vertical and horizontal conductances respectively, and  $\chi$  the iteration number.

We fixed the membrane potential at the value  $-70$  mV and  $L$  at 500. Fig. 3 shows the membrane conductance ( $g_m$ ) for four open probability ( $P_K$ ) and different open probabilities: ( $P_{Ca}$  and  $P_{Na}$ ).

We notice that for  $P_K = 0.1$  and  $0.15$ , the membrane conductance changes quickly from value 0 to 20 pS/cm<sup>2</sup>



**Figure 3.** Variation of the membrane conductance ( $g_m$ ) according to the probabilities  $P_K$ ,  $P_{Na}$  and  $P_{Ca}$ . The percolation threshold is reached when the  $g_m$  passes from 0 to 0.1 pS/cm<sup>2</sup>, for  $P_K = 0.2$ ,  $P_{Na} = 0.45$  and  $P_{Ca} = 0.35$ .



**Figure 4.** Voltage-clamp simulation. **A.** 3 s depolarisation from  $-70$  to  $0$  mV. **B.** Variation of the  $C_{Ca}$  intracellular concentration. **C.** Compartment of the current membrane for  $C_{Na} = 10$  mmol/l (solid line) and  $C_{Na} = 20$  mmol/l (dot line).

and to  $30$  pS/cm<sup>2</sup>, respectively. For  $P_K = 0.3$ , the membrane conductance is non null for all probabilities. These two behaviours do not describe the phase transition in the vicinity of the percolation threshold (Stauffer and Aharony 1991). On the other hand, for  $P_K = 0.2$ , the membrane conductance passes from  $0$  to  $0.1$  pS/cm<sup>2</sup> with  $P_{Ca} = 0.35$  and  $P_{Na} = 0.45$ , this is the percolation threshold of our modelisation. The network is composed of 20%  $K_{Ca}$ , 35%  $Ca_v$  and 45%  $Na_v$  channels.

### Voltage clamp technique

In this part, we studied the variation of the intracellular  $Ca^{2+}$  concentration and the membrane current ( $I_m$ ) resulting from a 3 s depolarization to  $0$  mV from a holding potential of  $-70$  mV (Fig. 4), and for two intracellular sodium concentrations  $10$  and  $20$  mmol/l.

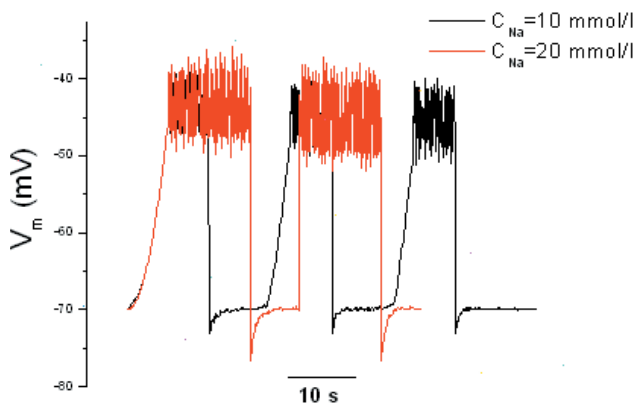
We notice on Fig. 4B that the depolarizing voltage from  $-70$  mV increases the  $C_{Ca}$ , reflecting Ca entry through  $Ca_v-L$  (L-type Ca channel) and NCX (we did not take into account the plasma membrane  $Ca^{2+}$ -ATPase pump function). The exchanger has a stoichiometry of  $3 Na^+$  for  $1 Ca^{2+}$ , is electrogenic and displays a reversal potential at  $-20$  and  $-40$  mV for  $C_{Na}$  equal to  $10$  and  $20$  mmol/l, respectively (Herchuelz et al. 2002), so that  $V_m$  is greater than reversal potential and favors  $Ca^{2+}$  entry (outward NCX current). Repolarization promotes  $Ca^{2+}$  removal by the exchanger, so that  $C_{Ca}$  decreases when  $V_m$  becomes more negative (inward NCX current). The simulation shows that elevated concentrations of cytosolic  $Na^+$  induce a mode of activity that no longer requires allosteric  $Ca^{2+}$  activation (Condrescu and Reeves 2006; Urbanczyk et al. 2006). We announce that high  $C_{Na}$  (inhibition of the Na/K pump) (Despa et al. 2002) reduces the  $Ca^{2+}$  extrusion through NCX. The  $C_{Ca}$  oscillation at repolarisation is due to the balance between an outward and an inward NCX current (Weber et al. 2003).

The membrane current in Fig. 4C highlights the presence of the Na/Ca exchanger. We notice that during the depolarization, the peak inward current is shorter in  $20$  than in  $10$  mmol/l intracellular  $Na^+$  because of the increased outward NCX current (Zhou and Lipsius 1993; McCarron et al. 1994). During the depolarization, the outward current is smaller in  $20$  than in  $10$  mmol/l  $Na^+$ , because of the inactivation of the  $Ca^{2+}$  current (Findlay et al. 2008). What is important is that at repolarization, the outward tail current is larger in  $20$  than in  $10$  mmol/l  $Na^+$ , due to the activation of the  $K_{Ca}$  current, resulting from the increased  $C_{Ca}$ .

### Current clamp technique

We imposed  $I_m = 10$  pA in order to have a resting potential of  $-70$  mV and we modelised the change in membrane potential for two concentrations of  $Na_v$ :  $10$  and  $20$  mmol/l.

For the two intracellular  $Na^+$  concentrations, we note a periodic electrical activity, the active phases are named "bursting" (Fig. 5). The burst duration is reduced when the  $C_{Na}$  decreased. The shorter burst duration led to a reduction of the  $Ca^{2+}$  influx (Sherman et al. 1988; Sherman and Rinzel 1991). A hyperpolarization of  $3.08$  mV and of  $6.7$  mV is announced for  $10$  and  $20$  mmol/l intracellular  $Na^+$  concentrations. This is in agreement with the previous results of the



**Figure 5.** Comparison of burst of membrane potential for two intracellular  $\text{Na}^+$  concentrations:  $C_{\text{Na}} = 10$  mmol/l (in black) and  $C_{\text{Na}} = 20$  mmol/l (in red), with a post-hyperpolarisation of 3.08 and 6.7 mV, respectively.

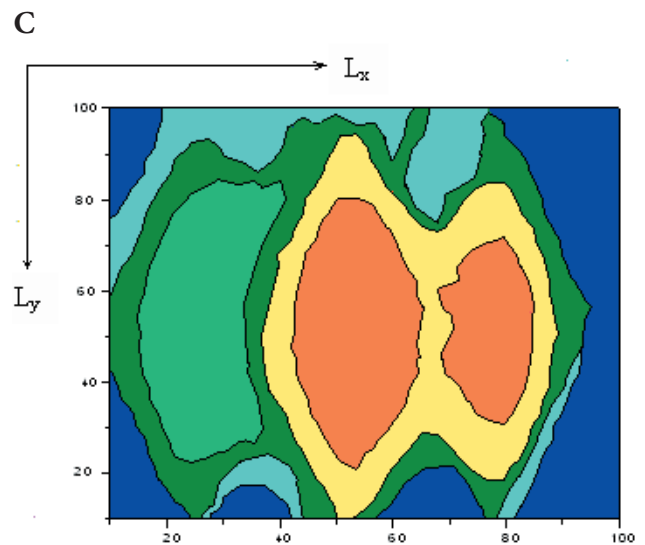
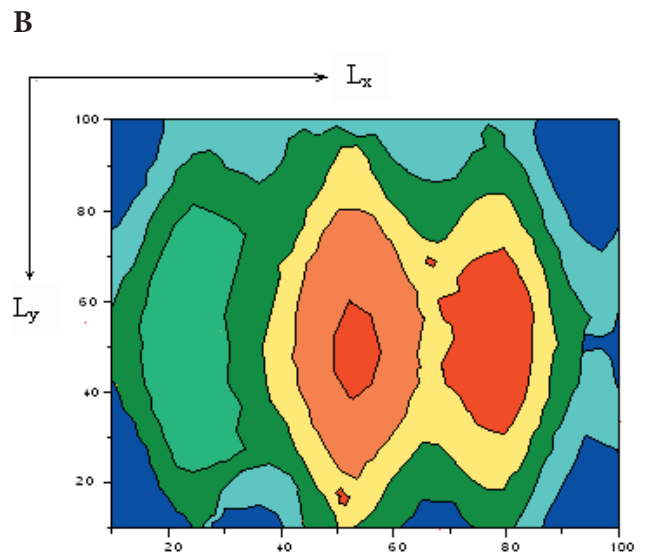
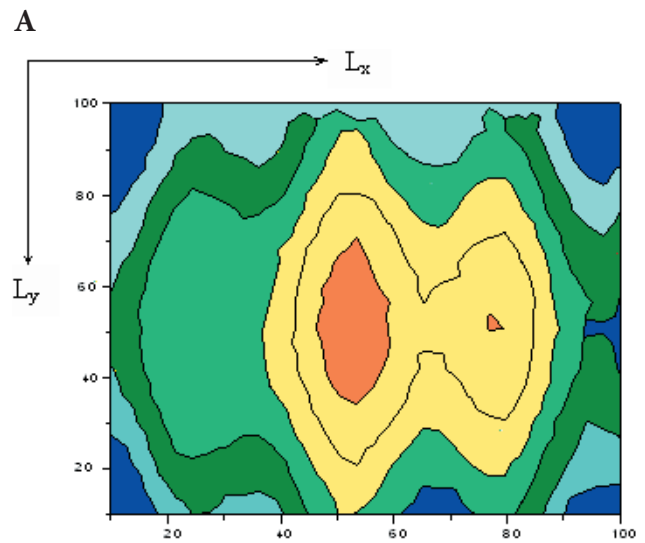
voltage clamp technique; the hyperpolarizing  $\text{K}^+$  current is more important when  $C_{\text{Na}}$  increases (Lee et al. 2002). For potentials higher than  $-45$  mV, we observed a repetitive bursting activity due to the spontaneously active channels (Reboreda et al. 2003).

*Distribution of the active channels*

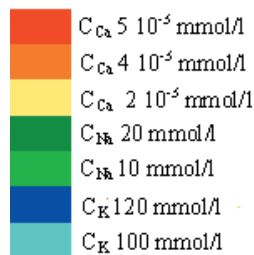
Since it is interesting to follow the evolution of the ion transfer through the plasma membrane, we studied the distribution of the active channels. For that, we represented the network structures of size  $100 \times 100$  for various ion concentrations (Fig. 6).

The  $\text{K}_{\text{Ca}}$  active channels are located in blue, the  $\text{Na}_v$  active channels in green and the  $\text{Ca}_v$  active channels in red.

We notice that for the Fig. 6, the concentrations in  $\text{K}_{\text{Ca}}$  are distributed on the four edges of the network that is in good



**Figure 6.** Temporal distribution of the active channels for a network  $100 \times 100$  with various ion concentrations. For the same ion, the colour is in range to indicate the areas moreover at least concentrated. The  $\text{K}_{\text{Ca}}$  active channels are located in blue, the  $\text{Na}_v$  active channels in green and the  $\text{Ca}_v$  active channels in red.  $t = 4, 5, 6$  s (A, B, C, respectively).



agreement with the FDM conditions. By exploring the three figures, we see that the concentration  $C_{Ca}$  increases from 2 to 5  $\mu\text{mol/l}$  in one second. This corresponds to a massive entry of  $\text{Ca}^{2+}$  ions. Then  $C_{Ca}$  returns to its basal concentration at  $t = 6$  s, characterised by an efflux of  $\text{Ca}^{2+}$  ions.

## Conclusion

By a simple electric model of the plasma membrane and by a theoretical approach based on the FDM and the percolation concept, we studied the role of the NCX in the regulation of the intracellular  $\text{Ca}^{2+}$  concentration and the membrane potential oscillations. It should be announced that the NCX and the plasma membrane  $\text{Ca}^{2+}$ -ATPase pump are two concurrent mechanisms for  $\text{Ca}^{2+}$  extrusion from the cell.

We detected the presence of an inward current, which is proportional to the intracellular  $\text{Na}^+$  concentration and an outward tail current due to the exit of  $\text{K}^+$  ions, which causes the hyperpolarization of the plasma membrane. The bursting duration is modified according to the  $\text{Na}^+$  intracellular concentration. Large activation corresponds to high  $C_{Na}$ . Temporal fractal structures of the active channel distributions enabled us to follow the diffusion of the  $\text{Ca}^{2+}$  intracellular ions. We compared these results with those of Espinosa Leon (Chouabe et al. 1997) and Zhengyi Wang (Wang et al. 2001) for the study of cardiac hypertrophy and David Gall (Gall et al. 1999) on the pancreatic  $\beta$ -cell. These works showed that the NCX is well implied in the lengthening of the action potential. The combination of the percolation and the FDM gave results in perfect agreement with the experimental results.

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